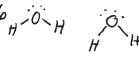
Succinate dehydrogenase catalyzes the conversion of succinate to fumarate.

1. Draw succinate and determine the number of valence electrons.

2. Draw fumarate and determine the number of valence electrons.

$$C - C = C - C$$

- 3. Is the conversion of succinate to fumarate an oxidation or reduction reaction? ολ. δαθίση
- 4. How many electrons will be carried by succinate dehydrogenase's redox cofactor (FAD)? 2
- 5. Balance the following reaction:



6. How many Cytochrome C (Fe²⁺) are required by Complex IV to process one O₂?